Disk-Shaped Amperometric Enzymatic Biosensor for in Vivo Detection of D-serine

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ABSTRACT: At the synapse, D-serine is an endogenous co-agonist for the N-methyl-D-aspartate receptor (NMDAR). It plays an important role in synaptic transmission and plasticity and has also been linked to several pathological diseases such as schizophrenia and Huntington’s. The quantification of local changes in D-serine concentration is essential to further understanding these processes. We report herein the development of a disk-shaped amperometric enzymatic biosensor for detection of D-serine based on a 25 μm diameter platinum disk microelectrode with an electrodeposited poly-m-phenylenediamine (PPD) layer and an R. gracilis D-amino acid oxidase (RgDAAO) layer. The disk-shaped D-serine biosensor is 1−5 orders of magnitude smaller than previously reported probes and exhibits a sensitivity of 276 μA cm⁻² mM⁻¹ with an in vitro detection limit of 0.6 μM. We demonstrate its usefulness for in vivo applications by measuring the release of endogenous D-serine in the brain of Xenopus laevis tadpoles.

A strocyte-derived D-serine is an endogenous co-agonist for N-methyl-D-aspartate receptors (NMDARs) at synapses in many regions of the mature brain, including the cerebral cortex, hippocampus, and cerebellum.1,2 NMDARs, a member of the glutamate receptor family, play an especially important role in basal synaptic transmission and as a critical mediator of many forms of synaptic plasticity.3,4 They have also been important clinical therapeutic targets in psychiatric disorders such as schizophrenia and depression,5,6 as well as neurodegenerative disorders like Huntington’s disease and amyotrophic lateral sclerosis (ALS).8,9 Activation of the NMDAR requires concurrent binding of its primary ligand glutamate, which is the most common excitatory neurotransmitter in the vertebrate central nervous system (CNS),10 at a selective glutamate binding site,11 together with binding of a co-agonist, either glycine or D-serine, at the so-called glycine binding site. Thus, astrocyte release of D-serine can modulate excitatory neuronal transmission and synaptic plasticity.12

In contrast to transmitter release by neurons, which has been extensively studied and characterized at the level of the underlying molecular machinery, the mechanisms responsible for astrocyte release of D-serine remain controversial and are less well understood. One reason for this is the fact that “gliotransmission” is thought mainly to exert modulatory influences on neuronal communication and to be less temporally restricted than the rapid vesicular neurotransmission between synaptic pairs of neurons. In contrast to electrically excitable neurons, glial cells do not exhibit a rapid depolarization of their membrane potential in response to neurochemical stimulation that can be measured using microelectrodes. However, a number of studies using calcium-sensitive fluorescent dyes have revealed cell-wide calcium transients in glial astrocytes which may initiate the release of gliotransmitters like D-serine.13,14 Moreover, rapid propagation of calcium elevations between astrocytes in culture has been observed, demonstrating the ability to communicate with each other through a network of gap junctional connections.15 Recent calcium imaging studies further suggest that astrocytes may exhibit very local calcium transients that can influence synaptic efficacy at nearby neuronal contacts.16,17 Thus, improving the ability to precisely localize extracellular changes in D-serine concentration at the cellular or even...
subcellular level constitutes a critical step toward better characterizing this fundamental aspect of neuron–glia communication.

Previous efforts to measure D-serine levels in situ have relied primarily on local sampling by microdialysis. Highly specific detection of D-serine in perfusion samples from the brain can be obtained using high-performance liquid chromatography (HPLC) or by fluorescence-based capillary electrophoresis (CE) of fluorescently derivatized distillates of cerebrospinal fluid (CSF). In combination with implantable microdialysis probes, limited spatial and temporal information can be obtained by these approaches. Several reports have shown that significantly improved spatiotemporal resolution can be provided by the use of enzymatic amperometric biosensors. One method of detection of D-serine using such probes is based on the production of H2O2 during the enzymatic degradation of D-serine by D-aminooxy acid oxidase (DAO). These biosensors can be fabricated small enough to insert directly into brain tissue of interest. Because D-amino acids, with the exception of D-amino acids, are extremely rare in the CNS, this method offers reasonable selectivity and spatial accuracy limited only by the geometry of the functionalized probe tip itself. Previously described D-serine biosensors have proven useful for in vivo applications, but have been based on functionalized tips too large to allow measurements on the scale of individual cells. Here we report the development of a 25 μm diameter disk amperometric enzymatic biosensor that is 1–5 orders of magnitude smaller than previously reported probes and which displays increased sensitivity for D-serine detection. We further demonstrate its usefulness for in vivo applications by measuring the unprecedented evoked release of endogenous D-serine in the brain of stage 48 albino Xenopus laevis tadpoles.

**EXPERIMENTAL SECTION**

**Chemicals.** Ascorbic acid, dopamine, D-serine, ferrocene-methanol (FcCH2OH), hydrogen peroxide, L-glutamate, m-phenylenediamine, ethyl 3-aminobenzoate methanesulfonate (MS222), and serotonin were purchased from Sigma-Aldrich (Oakville, Canada). Calcium chloride, calcium nitrate, glucose, HEPES, magnesium chloride, magnesium sulfate, potassium chloride, sodium chloride, and sodium bicarbonate were purchased from VWR (Mississauga, Canada). Tetrodotoxin (TTX), cyclohexamide, and 2-amino-3-(hydroxy-5-methyl-isoxazol-4-yl) propanoic acid (AMPA) were purchased from R&D Systems, Inc. (Minneapolis, U.S.A.). Other chemical reagents were purchased from Fisher Scientific (Ottawa, Canada), unless otherwise noted.

**Enzyme Preparation.** Recombinant* R. gracilis d-amino acid oxidase (RgDAOO, EC 1.4.3.3) was overexpressed in* E. coli cells and purified to homogeneity as previously reported. The final enzyme solution was concentrated to 50 mg/mL protein in PBS (0.01 M, pH 7.4) with 1% glycerol and 25 mg/mL bovine serum albumin (BSA); pure RgDAOO had a specific activity of 75 ± 7 U/mg protein on D-serine as substrate based on amperometric assay.

**Electrochemical Measurements.** Electrochemical measurements were performed using either an Electrochemical Probe Scanner 3 (Heka Elektronik, Lambrecht, Germany) or an Axopatch 200B amplifier (Molecular Devices, Sunnyvale, U.S.A.). All potentials were recorded relative to a chloridized silver wire (in house) quasi-reference electrode. In vitro calibrations in 0–50 μM D-serine in PBS (0.01 M, pH 7.4) were obtained using chronoamperometry (10 min, 500 mV) with a three-electrode setup (Pt wire counter electrode). In vivo calibrations were also performed using chronoamperometry after insertion into the optic tectum of tadpoles. In vivo standard solutions were made in external artificial cerebrospinal fluid (ACSF) containing 115 mM NaCl, 2 mM KCl, 3 mM CaCl2, 3 mM MgCl2, 5 mM HEPES, 10 mM glucose, 1 μM TTX (to prevent neuronal activity), at pH 7.4. All in vivo measurements were performed using a two-electrode system.

**Preparation of Biosensor Backbone: 25 μm Pt Disk Microelectrode.** Microelectrodes were fabricated by initially pulling a soda-lime glass capillary (Hilgenberg GmbH, Malsfeld, Germany) using a P-2000 micropipette puller (Sutter Instruments, Novato, CA). A 25 μm diameter Pt wire (Goodfellow, Huntington, U.K.) was then inserted into the capillary and sealed using a PC-10 vertical pipette puller (Narishige, Japan) under vacuum. The Pt wire was electrically connected to a copper wire using silver epoxy (H20E, Epo-Tek, Billerica, U.S.A.), which was then cured at 150 °C for 30 min. To increase mechanical stability, the microelectrode was inserted into a larger borosilicate glass capillary (Sutter Instruments, Novata, U.S.A.), with an overlap of 1.5 cm. A gold connector (HEKA Elektronik, Lambrecht, Germany) was soldered onto the copper wire, and junctions were sealed with epoxy glue (Henkel Canada, Mississauga, Canada). The microelectrode tip was mechanically polished (400 rpm, 4000 grit Silicon carbide grinding paper, 15 min) using a TegraPol-25 grinder/polisher (Struers Ltd., Mississauga, Canada) until the Pt wire was exposed as a disk. The microelectrode was then washed in 18.2 MΩ water, 70% ethanol, and acetone in preparation for subsequent steps. The diameter of the microelectrode was characterized using cyclic voltammetry (3 cycles, –100 mV to +500 mV, 5 mV s–1) in 1 mM FeCl2/Fe(OH)2 (0.1 M KCl). The RG of the microelectrode, defined as the ratio of the radius of the entire microelectrode (glass + Pt wire) to that of the Pt metal wire, was determined by optical microscopy using a customized Axio Vert.A1 inverted microscope (Zeiss, Oberkochen, Germany).

**Addition of Polymer and Enzyme Layer.** A layer of 100 mM poly-meta-phenylenediamine (PPD) in PBS (0.01 M, pH 7.4) was electropolymorized onto the Pt surface using cyclic voltammetry (15 cycles, 0 to +1000 mV, 100 mV s–1). The PPD-modified microelectrode was then immersed tip-up in a 0.5 μL droplet of final enzyme solution, formed at the aperture of a micropipette tip. The enzyme meniscus was allowed to evaporate, leaving approximately 2 μL of adsorbed enzyme on the PPD-modified microelectrode. The biosensor was then placed in a sealed glass chamber containing 10 mL of glutaraldehyde (50% v/v in H2O2), as a source of vapor, to crosslink the RGDAOO for 10 min.

Null biosensors were produced using the exact same procedure described above, with the exception of the active RGDAOO layer. Instead, the PPD-modified microelectrode was immersed in a solution of 400 mg/mL BSA in PBS (0.01 M, pH 7.4) and crosslinked using glutaraldehyde.

**Tadpole Preparation and In Vivo Measurements.** Brains from stage 48 albino Xenopus laevis tadpoles were dissected out following anesthesia by immersion in 0.02% MS222 0.1X MBS-H (Modified Barth’s Saline with HEPES) rearing solution and placed in ACSF. The rearing solution was composed of 880 μM NaCl, 10 μM KCl, 24 μM NaHCO3, 8.2 μM MgSO4·(6H2O), 3.3 μM Ca(NO3)2·(4H2O), 4.1 μM CaCl2·(2H2O), and 100 μM HEPES, adjusted to pH 7.4. To expose the ventricular and pial surfaces of the optic tectum,Brains from stage 48 albino Xenopus laevis tadpoles were dissected out following anesthesia by immersion in 0.02% MS222 0.1X MBS-H (Modified Barth’s Saline with HEPES) rearing solution and placed in ACSF. The rearing solution was composed of 880 μM NaCl, 10 μM KCl, 24 μM NaHCO3, 8.2 μM MgSO4·(6H2O), 3.3 μM Ca(NO3)2·(4H2O), 4.1 μM CaCl2·(2H2O), and 100 μM HEPES, adjusted to pH 7.4. To expose the ventricular and pial surfaces of the optic tectum,
enzymatic degradation of D-serine. Upon binding of D-serine, the D-serine oxidase (DAAO) enzyme layer (light gray) to react with D-serine, a reductive half-reaction occurs, leading to the production of hydroxypyruvate and ammonia. Molecular oxygen oxidizes FADH$_2$ back to FAD, producing equimolar H$_2$O$_2$, which then diffuses through the interface blocking PPD layer and is oxidized at the biased Pt surface. (B) Schematic representation of the in vivo measurement environment within the brain (not to scale). The disk-shaped biosensor is used to measure evoked release of D-serine from astrocytes.

Figure 1. Principle of D-serine detection using a disk-shaped amperometric enzymatic biosensor. (A) D-serine reacts with immobilized RgDAAO, degrading into hydroxypyruvate and ammonia. Molecular oxygen oxidizes FADH$_2$ back to FAD, producing equimolar H$_2$O$_2$, which then diffuses through the interface blocking PPD layer and is oxidized at the biased Pt surface. (B) Schematic representation of the in vivo measurement environment within the brain (not to scale). The disk-shaped biosensor is used to measure evoked release of D-serine from astrocytes.

RESULTS AND DISCUSSION

Figure 1. Principle of D-serine detection using a disk-shaped amperometric enzymatic biosensor. (A) D-serine reacts with immobilized RgDAAO, degrading into hydroxypyruvate and ammonia. Molecular oxygen oxidizes FADH$_2$ back to FAD, producing equimolar H$_2$O$_2$, which then diffuses through the interference blocking PPD layer and is oxidized at the biased Pt surface. (B) Schematic representation of the in vivo measurement environment within the brain (not to scale). The disk-shaped biosensor is used to measure evoked release of D-serine from astrocytes.

Brains were filed down the midline and pinned to a piece of Sylgard, which was submerged in a recording chamber perfused with fresh ACSF.

A Narishige micromanipulator was used to position the biosensor on the ventricular surface of the brain adjacent to the tectal cell layer. The brain was then allowed to stabilize for 20 min in this configuration. D-serine release was stimulated by perfusing ACSF containing 100 μM AMPA with 50 μM cyclothiazide to activate AMPA-type glutamate receptors and to block receptor desensitization, respectively.

All animal experiments were approved by the Montreal Neurological Institute Animal Care Committee.

Data Analysis. Data are presented as mean ± standard error of the mean (S.E.M.). Oxidation current values are presented as blank corrected currents. The number of data points is defined as n. The limit of detection (LOD) is defined as 3 times the standard deviation of the blank divided by the slope of the regression line. The limit of quantitation (LOQ) is defined as 10 times the standard deviation of the blank divided by the slope of the regression line. Data sets were analyzed using either Excel 2010 (Microsoft Office) or Matlab R2013a (Mathworks, Natick, U.S.A.).

Principle of Detection of D-Serine Release in the Brain. An enlarged schematic of the biosensor tip is shown in Figure 1A. It consists of three components including an RgDAAO enzyme layer (light gray) to react with D-serine, a PPD polymer layer (dark gray) to block out large electroactive interferences, and a platinum surface (black) used for the oxidation of H$_2$O$_2$. Detection of D-serine using the biosensor relies on the stoichiometric production of H$_2$O$_2$ during the enzymatic degradation of D-serine. Upon binding of D-serine with RgDAAO, a reductive half-reaction occurs, leading to the production of hydroxypyruvate and ammonia, along with reduced flavin adenine dinucleotide (FADH$_2$). Subsequently, an oxidative half-reaction with molecular oxygen occurs to produce FAD and produces an equimolar amount of H$_2$O$_2$ (Scheme 1). H$_2$O$_2$ then diffuses through the PPD layer and is oxidized at the biased Pt surface (Scheme 2).

Using this detection principle in combination with D-serine calibration methods, the local concentration of D-serine can be determined in vivo.

Figure 1B shows a schematic of the in vivo measurement environment (not to scale), where the disk-shaped biosensor is inserted into the optic tectum of the tadpole. The proposed release of D-serine consists of four steps, as has been previously described.

1. An action potential reaches the axon terminal of the presynaptic neuron, (2) triggering the release of the neurotransmitter, glutamate. (3) Glutamate, which is an agonist for both AMPA and NMDA receptors, causes AMPA receptors to activate. Activation of the AMPA receptors present on the astrocyte triggers a cascade leading to the release of D-serine. (4) Both glutamate and D-serine, which are co-agonists of the NMDAR at synapses, bind and activate it, allowing cation influx which depolarizes the postsynaptic neuron. Detection of local release of D-serine during this process requires a biosensor that is both highly sensitive and geometrically appropriate to allow sampling with precise spatial resolution. For this reason, the disk-shaped biosensor was required.

Biosensor Geometry. Previously reported D-serine biosensors capable of performing in vivo measurements mostly display a cylindrical geometry (e.g., 25 μm diameter and 150 μm long) several times larger than the brain cells targeted in tadpoles. The cell body of a large cortical pyramidal neuron is approximately 25 μm in diameter. The entire brain structure studied herein, the optic tectum in a stage 48 Xenopus laevis tadpole, measures less than 200 μm along its mediolateral axis. Although these cylindrical biosensors have a large surface area and relatively good sensitivity, their size and geometry make it difficult to decipher cell body response versus peripheral processes, specific layers of the tectum, or to determine if the observed response originated within the optic tectum. The developed disk-shaped biosensor provides a useful solution to this problem because it can be accurately positioned within the optic tectum, effectively acting as a spatially targeted probe.

To avoid bending previously reported in cylindrical biosensors and improve manipulation in vivo, the configuration
of the disk-shaped biosensor seals the entire metal wire in a glass matrix, as shown in Figure 2A. The side view optical micrograph shows a 25 μm diameter Pt wire surrounded by a thin glass sheath. The RG of the Pt microelectrode was 2.6 ± 0.1 (n = 15) as determined from optical microscopy (see SI; Figure S1). Biosensors fabricated from microelectrodes having such RGs were mechanically stable while retaining the desired spatial resolution for in vivo measurements in tadpoles. Successful exposure of the Pt wire during mechanical polishing is demonstrated by the top view optical micrograph in Figure 2A, which shows a centered Pt wire surrounded by a thin glass sheath. Following electropolimerization of PPD onto the Pt surface using cyclic voltammetry (Figure 2B), the polymer layer is not thick enough to be observed from the side view optical micrograph (Figure 2C). Successful electrodeposition of PPD is noticeable from the top view optical micrograph (Figure 2C) based on the granular appearance of the microelectrode and is further confirmed by cyclic voltammetry using FeCH₂OH, where the initial faradic behavior of the bare Pt surface changed to an ohmic behavior following PPD electropolymerization (see SI; Figure S2). The RgDAAO enzyme layer is adsorbed over PPD by immersion in a stable meniscus of final enzyme solution formed at a micropipette (Figure 2D). Following immobilization of RgDAAO using glutaraldehyde vapors, a stable enzyme–PPD matrix is formed.38,39 The immobilized enzyme was yellow in color, meaning it was in its holoenzyme form. The full biosensor assembly is shown in Figure 2E. The diameter of the final disk-shaped biosensor (approximately 80 μm) is suitable for in vivo studies in the optic tectum of tadpoles. Practically, these sensors can be recycled easily by polishing for 5 min to regenerate the original 25 μm Pt backbone. A new biosensor can be fabricated by replenishing the surface chemistry within 1 h, greatly diminishing total fabrication time and decreasing the total fabrication costs.

**Biosensor Selectivity.** Insertion of a biased biosensor into living tissue severely complicates the sample matrix due to the presence of other oxidizable molecules, or interferences. Several studies have reported the use of polymer membranes to prevent biological interferences from reaching an electrode surface.30 More specifically, PPD has been shown to be particularly effective in excluding ascorbic acid, a highly desirable characteristic for this biosensor.41,42 PPD works using the size exclusion principle, whereby relatively large molecules such as ascorbic acid and dopamine are unable to diffuse through the polymer membrane, which remains permeable to H₂O₂, a small molecule.

In order to verify that oxidation currents recorded in vivo were in fact related to oxidation of H₂O₂, chronoamperometric measurements were performed without and with a PPD layer, in solutions containing physiological concentrations of several possible interfering molecules that could be oxidized at 500 mV. These molecules included ascorbic acid (AA), dopamine (DA), 3,4-dihydroxyphenylacetic acid (DOPAC), serotonin (5-HT), 5-hydroxyindoleacetic acid (5-HIAA), homovanillic acid (HVA), and glutamate (GLUT).43 The recorded oxidation currents of these interferences at a bare 25 μm Pt disk microelectrode (without PPD) are shown by black bars in Figure 3A. HVA shows a relatively low oxidation current, whereas glutamate produced no detectable current. Consequently, even at relatively high concentrations in the sample matrix of the brain, the contributions of these two molecules to subsequent electroactive interferences (Figure 3A, white bars). For AA, DA, 5-HT, 5-HIAA, and DOPAC, the observed percent decrease in oxidation current corresponded to 77.6 ± 0.7, 71 ± 2, 99.1 ± 0.5, 99.9 ± 0.1, and 92 ± 1, respectively (n = 5). These decreases confirm that the PPD layer can effectively block out electroactive interferences.

However, the physical presence of a PPD layer also hinders the diffusion of H₂O₂ toward the Pt surface. Consequently, it is important to verify that H₂O₂ can be detected at a sufficiently low detection limit. Calibration curves obtained in standard solutions of H₂O₂ (2.5–25 μM) without and with PPD showed

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**Figure 2.** Fabrication of a 25 μm diameter disk-shaped amperometric enzymatic biosensor. (A) 25 μm Pt disk microelectrode tip. Top and side view optical micrographs show a Pt core centered within a thin insulating glass sheath. (B) A PPD layer is electropolimerized onto the Pt surface using cyclic voltammetry. (C) Disk microelectrode tip with an electropolymerized PPD layer. Optical micrographs confirm successful addition of PPD. (D) RgDAAO is deposited using a micropipette. The enzyme layer is crosslinked using glutaraldehyde vapors, producing a bulb at the tip of the biosensor. (E) Complete biosensor assembly. Scale bars represent 50 μm.
a 51 ± 9% \( (n = 3) \) decrease in oxidation current (Figure 3B). The LOD of \( \text{H}_2\text{O}_2 \) with a PPD-modified microelectrode was 0.2 \( \mu \text{M} \), which confirms that although the permselective PPD layer slightly hinders its diffusion toward the Pt surface, it is still possible to detect \( \text{H}_2\text{O}_2 \).

**In Vitro Characterization of Biosensor.** The experimental disk-shaped biosensor was characterized in vitro using standard solutions of \( \text{D}-\text{serine} \) in PBS. Figure 4 shows a calibration curve for 2.5–50 \( \mu \text{M} \) \( \text{D}-\text{serine} \), which is a concentration range relevant for in vivo tadpole studies. For comparison purposes, a calibration curve was obtained for both the experimental disk-shaped biosensor and a commercially available cylindrical biosensor (DSER probe; Sarissa Biomedical). In order to correct for the different geometries of both biosensors, recorded oxidation currents were normalized for surface area. Consequently, the values displayed in Figure 4 are shown in terms of current density (\( \mu \text{A} \text{cm}^{-2} \)). The slopes of the regression lines shown demonstrate that although the commercial cylindrical biosensor has a 41-fold greater surface area, the experimental disk-shaped biosensor has a higher sensitivity to \( \text{D}-\text{serine} \). A comprehensive comparison with all previously reported \( \text{D}-\text{serine} \) biosensors is shown in Table 1. Although the experimental disk-shaped biosensor has a geometry 1–5 orders of magnitude smaller than other probes, it shows at minimum a 30% higher sensitivity. The improved biosensor sensitivity is likely related to its disk-shaped geometry, which maximizes the ratio of enzymatic to amperometric reaction surface areas as well as the enhanced specific activity of pure \( \text{RgDAAO} \ (75 ± 7 \text{ U/mg protein on \( \text{D}-\text{serine} \) as substrate}) \), as compared to other enzyme homologues (e.g., pig kidney \( \text{DAAO} \); 1.5 \text{ U/mg protein}).

**In Vivo Measurements in Tadpoles.** Following characterization and calibration of the experimental disk-shaped biosensor, chronoamperometric measurements were performed in the optic tectum of tadpoles \( (n = 5) \). A representative response is shown in Figure 5. After positioning in the optic tectum, a 500 mV potential was applied, and the biosensor was allowed a 20 min stabilization period in ACSF. Standard \( \text{D}-\text{serine} \) solutions in ACSF (0–30 \( \mu \text{M} \)) were then flowed into the sample chamber. Measurement of \( \text{D}-\text{serine} \) concentration standards washed on in this configuration produced a linear response (Figure 5 inset) with a limit of detection of 0.6 ± 0.1 \( \mu \text{M} \) and a limit of quantitation of 2.1 ± 0.2 \( \mu \text{M} \) \( (n = 3) \). The sensitivity of the disk-shaped biosensor in vivo was 279 ± 21 \( \mu \text{A cm}^{-2} \text{mmol}^{-1} \text{min}^{-1} \) \( (n = 3) \), demonstrating the functionality of the biosensor within both in vitro and in vivo environments. Following in vivo calibration, the biosensor remained inside the optic tectum until the current response returned back to the original steady-state level in ACSF (~10 min). The AMPA-type glutamate receptor agonist AMPA, along with cyclothiazide to block receptor desensitization, was then added to the perfusion, producing an increase in current measured at the biosensor, which reached a plateau within 2 min. This increase in current represents the release of endogenous \( \text{D}-\text{serine} \) in response to receptor activation and constitutes the first time to our knowledge that release of endogenous \( \text{D}-\text{serine} \) has been

![Figure 3](Image 366x590 to 523x749)

**Figure 3.** Permeability of the biosensor before and after electro-deposition of a permselective PPD layer. (A) Biosensor response toward ascorbic acid (AA), dopamine (DA), 3,4-dihydroxyphenylacetic acid (DOPAC), serotonin (5-HT), 5-hydroxyindoleacetic acid (5-HIAA), homovanillic acid (HVA), and glutamate (GLUT), without (black) and with (white) PPD layer. (B) Calibration curves for 2.5 \( \mu \text{M} \) \( \text{D}-\text{serine} \) in PBS. Linear regression equation without PPD: \( J = 6 \times C (\mu \text{M}) - 11; R^2 = 0.995 \); LOD = 0.8 ± 0.2 \( \mu \text{M} \); LOQ = 0.7 ± 0.1 \( \mu \text{M} \). Currents were measured at constant potential (500 mV).

![Figure 4](Image 100x408 to 261x749)

**Figure 4.** In vitro characterization and comparison of \( \text{D}-\text{serine} \) biosensor. Response toward 0–50 \( \mu \text{M} \) \( \text{D}-\text{serine} \) was measured using a commercially available cylindrical biosensor (dashed line; \( n = 5 \)) and the experimental disk-shaped biosensor (solid line; \( n = 3 \)). Linear regression equation for the experimental disk-shaped biosensor: \( J = 0.3 \times C (\mu \text{M}) - 0.2; R^2 = 0.995 \); LOD = 0.6 ± 0.1 \( \mu \text{M} \); LOQ = 2.1 ± 0.2 \( \mu \text{M} \). Linear regression equation for the commercial cylindrical biosensor: \( J = 0.1 \times C (\mu \text{M}) + 0.1; R^2 = 0.995 \); LOD = 0.007 ± 0.002 \( \mu \text{M} \); LOQ = 0.02 ± 0.01 \( \mu \text{M} \). Currents were measured at constant potential (500 mV).
Article

Table 1. Comparison of Biosensor Properties with Previously Reported D-serine Biosensors

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<th>source</th>
<th>geometry** (μm)</th>
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**Diameter × Height. bThe surface area ratio is defined as the ratio of the electroactive surface area of other biosensors to that of the experimental biosensor.

![Image of in vivo detection of D-serine inside the optic tectum of a Xenopus laevis tadpole.](Image)

Figure 5. In vivo detection of D-serine inside the optic tectum of a Xenopus laevis tadpole. Standard solutions of D-serine were flowed into the measurement chamber to calibrate the biosensor in vivo (t = 20–30 min). Perfusion of AMPA into the chamber (t = 47 min) stimulated release of endogenous D-serine, which was then measured locally. Inset: calibration curve obtained in vivo for 0–30 μM D-serine (n = 3). Linear regression equation: I (pA) = 1.4 × C (μM)−0.6; R² = 0.995; LOD = 0.6 ± 0.1 μM; LOQ = 2.1 ± 0.3 μM. Currents were measured at constant potential (500 mV).

The current response obtained using the disk-shaped experimental D-serine biosensor was in fact related to oxidation of D-serine and not other electroactive species.

## CONCLUSIONS

We have successfully designed and fabricated a fully functional disk-shaped amperometric enzymatic biosensor for detection of D-serine, usable in both in vitro and in vivo environments. Its geometry allows for increased mechanical robustness and high spatial resolution. Furthermore, the experimental disk-shaped biosensor exhibited a higher sensitivity to D-serine than any other previously reported amperometric probe.

Although its use in vivo was demonstrated here in the optic tectum of tadpoles, this biosensor could be readily adapted to applications for the study of D-serine contributions to neuroplasticity in acute brain slices or in vivo. Its small size allows for spatially precise measurements not possible using currently available amperometric probes. For example, it could be used to measure D-serine availability and release in different layers of the cerebral cortex. Its small profile would also make it suitable for simultaneous calcium imaging to better assess the temporal relationship of glial calcium transients to glutamatergic release and the signaling cascades contributing to D-serine release. Potential applications for dissociated cultured cells might include a systematic mapping of the subcellular locations of presumptive sites of D-serine release from astrocytes and neurons. Overall, this biosensor provides a useful tool for detection of D-serine in any experimental situation where an amperometric probe could be applied.

## ASSOCIATED CONTENT

### Supporting information

Additional information as noted in the text. This material is available free of charge via the Internet at http://pubs.acs.org.

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D.P and A.K contributed equally to this work. The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.
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